

Series : QQCRR/2



SET-2

प्रश्न-पत्र कोड 31/2/2
Q.P. Code

रोल नं.

Roll No.

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परीक्षार्थी प्रश्न-पत्र कोड को उत्तर-पुस्तिका के मुख-पृष्ठ पर अवश्य लिखें।

Candidates must write the Q.P. Code on the title page of the answer-book.

- कृपया जाँच कर लें कि इस प्रश्न-पत्र में मुद्रित पृष्ठ 12 हैं।
- प्रश्न-पत्र में दाहिने हाथ की ओर दिए गए प्रश्न-पत्र कोड को छात्र उत्तर-पुस्तिका के मुख-पृष्ठ पर लिखें।
- कृपया जाँच कर लें कि इस प्रश्न-पत्र में 15 प्रश्न हैं।
- कृपया प्रश्न का उत्तर लिखना शुरू करने से पहले, उत्तर-पुस्तिका में प्रश्न का क्रमांक अवश्य लिखें।
- इस प्रश्न-पत्र को पढ़ने के लिए 15 मिनट का समय दिया गया है। प्रश्न-पत्र का वितरण पूर्वाह्न में 10.15 बजे किया जाएगा। 10.15 बजे से 10.30 बजे तक छात्र केवल प्रश्न-पत्र को पढ़ेंगे और इस अवधि के दौरान वे उत्तर-पुस्तिका पर कोई उत्तर नहीं लिखेंगे।
- Please check that this question paper contains 12 printed pages.
- Q.P. Code given on the right hand side of the question paper should be written on the title page of the answer-book by the candidate.
- Please check that this question paper contains 15 questions.
- **Please write down the Serial Number of the question in the answer-book before attempting it.**
- 15 minute time has been allotted to read this question paper. The question paper will be distributed at 10.15 a.m. From 10.15 a.m. to 10.30 a.m., the candidates will read the question paper only and will not write any answer on the answer-book during this period.



विज्ञान SCIENCE

निर्धारित समय : 2 घण्टे

Time allowed : 2 hours

अधिकतम अंक : 40

Maximum Marks : 40

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P.T.O.

General Instructions :

Read the following instructions carefully and strictly follow them.

- (i) This question paper contains **15** questions. **All** questions are compulsory.
- (ii) This question paper is divided into **three** Sections viz. Section **A, B** and **C**.
- (iii) Section **A** – Question numbers **1** to **7** are short answer type questions. Each question carries **two** marks.
- (iv) Section **B** – Question numbers **8** to **13** are also short answer type questions. Each question carries **three** marks.
- (v) Section **C** – Question numbers **14** and **15** are case based questions. Each question carries **four** marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

SECTION – A

1. In the following food chain, only 2J of energy was available to the peacocks. How much energy would have been present in Grass ? Justify your answer. 2

GRASS → GRASS HOPPER → FROG → SNAKE → PEACOCK

OR

- (a) What is meant by garbage ? List two classes into which garbage is classified.
 - (b) What do we actually mean when we say that the “enzymes are specific in their action” ?
2. (a) Name the poles P, Q, R and S of the magnets in the following figures ‘a’ and ‘b’ : $\frac{1}{2} + \frac{1}{2} + 1 = 2$

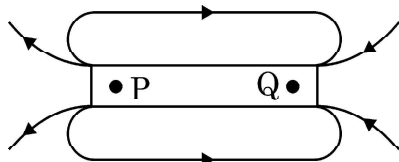


Figure ‘a’

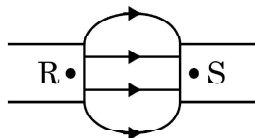


Figure ‘b’

- (b) State the inference drawn about the direction of the magnetic field lines on the basis of these diagrams.

OR

When is the force experienced by a current – carrying straight conductor placed in a uniform magnetic field. $1 + 1 = 2$

- (i) Maximum ;
- (ii) Minimum ?

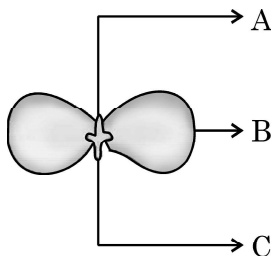


3. Name the reproductive parts of an angiosperm. Where are these parts located ? Explain the structure of its male reproductive part. 2

OR

What is puberty ? Mention any two changes that are common to both boys and girls in early teenage years.

4. (a) Name the reproductive and non-reproductive parts of bread mould (Rhizopus).
 (b) List any two advantages of vegetative propagation. 2
5. In the following figure showing a germinating gram seed, name the parts labelled as A, B and C : $\frac{1}{2} \times 4 = 2$



Why is part 'B' considered to be important during germination ?

6. A part of modern periodic table is given below. On its basis, answer the following questions : $\frac{1}{2} \times 4 = 2$

Group No. →	1	2	13	14	15	16	17	18
Period ↓								
2		A				B		
3	E			D			F	C

- (a) Write the molecular formula of the compound formed by the combination :
 (i) A and F (ii) E and B
- (b) Which of the element is a
 (i) Noble gas (ii) Metalloid
7. Write the chemical formula of two consecutive homologous of organic compounds having functional group -OH . What happens to the (i) boiling point and (ii) solubility of organic compounds of a homologous series as the molecular mass increases. $\frac{1}{2} \times 4 = 2$

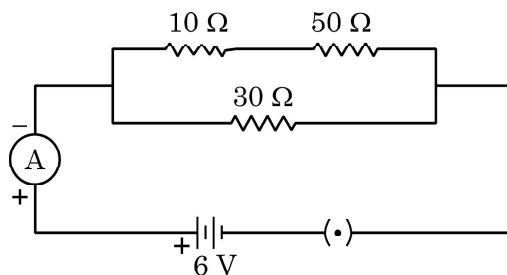


SECTION – B

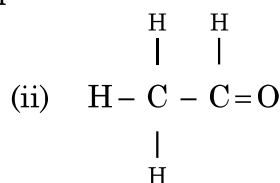
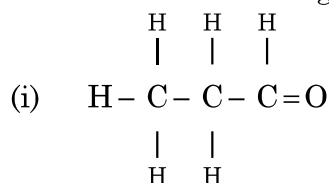
8. (a) We do not clean ponds or lakes, but an aquarium needs to be cleaned regularly. Why ? 1+2
 (b) Why is ozone layer getting depleted at the higher levels of the atmosphere ? Mention one harmful effect caused by its depletion.
9. (a) List the factors on which the resistance of a uniform cylindrical conductor of a given material depends. 2+1
 (b) The resistance of a wire of 0.01 cm radius is $10\ \Omega$. If the resistivity of the wire is $50 \times 10^{-8}\ \Omega\text{ m}$, find the length of this wire.

OR

- (a) What is the meaning of electric power of an electrical device ? Write its SI unit. 1½
 (b) An electric kettle of 2kW is used for 2h. Calculate the energy consumed in
 (i) kilowatt hour and
 (ii) joules. 1½
10. In the given circuit determine the value of : 3
 (i) total resistance of the circuit
 (ii) current flowing through the ammeter.



11. A green stemmed tomato plant denoted by (GG) is crossed with a tomato plant with purple stem denoted by (gg). 3
 (i) What colour of the stem would you expect in their F₁ progeny ?
 (ii) In what ratio would you find the green and purple coloured stem in plants of F₂ progeny ?
 (iii) What conclusion can be drawn for the above observations ?
12. Consider the following organic compounds : 3



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- (a) Name the functional group present in their compounds.
 (b) Write the general formula for the compounds of this functional group.
 (c) State the relationship between these compounds and draw the structure of any other compound having similar functional group.

OR

- (a) Draw the electron dot structure for ethyne. 1+2 = 3
 (b) List two differences between the properties exhibited by covalent compounds and ionic compounds.
13. (a) State Newland Law of Octaves. 1+1+ $\frac{1}{2}$ + $\frac{1}{2}$ = 3
 (b) With an example, explain Dobereiner's Triads.
 (c) List one limitation each of both the attempts mentioned in 'a' & 'b'.

SECTION - C

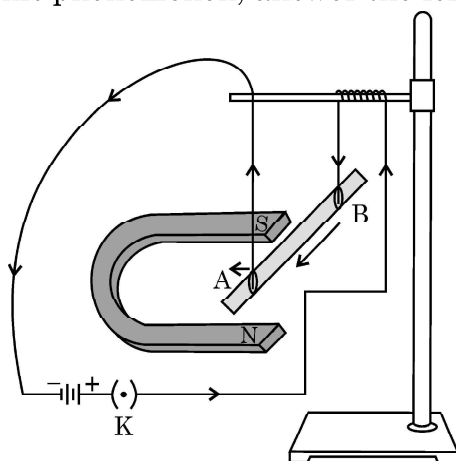
This section has 02 case based questions (14 and 15).

Each case is followed by 03 sub-questions (a, b and c).

Part (a) and (b) are compulsory. However an internal choice has been provided in Part (c).

14. A student was asked to perform an experiment to study the force on a current carrying conductor in a magnetic field. He took a small aluminum rod AB, a strong horse shoe magnet, some connecting wires, a battery and a switch and connected them as shown. He observed that on passing current, the rod gets displaced. On reversing the direction of current, the direction of displacement also gets reversed. On the basis of your understanding of this phenomenon, answer the following questions :

4



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- (a) Why does the rod get displaced on passing current through it ?
- (b) State the rule that determines the direction of the force on the conductor AB.
- (c) (i) In the above experimented set up, when current is passed through the rod, it gets displaced towards the left. What will happen to the displacement if the polarity of the magnet and the direction of current both are reversed ?
- (ii) Name any two devices that use current carrying conductors and magnetic field.

OR

Draw the pattern of magnetic field lines produced around a current carrying straight conductor held vertically on a horizontal cardboard. Indicate the direction of the field lines as well as the direction of current flowing through the conductor.

15. Mendel blended his knowledge of Science and mathematics to keep the count of the individuals exhibiting a particular trait in each generation. He observed a number of contrasting visible characters controlled in pea plants in a field. He conducted many experiments to arrive at the laws of inheritance.

*

- (a) If only one pair of contrasting characters like tall and short plants is taken, plants obtained in F1 generation are not of medium height. Why ?
- (b) Name the recessive traits in above case.
- (c) Mention the type of the new combinations of plants obtained in F2 progeny along with their ratio, if F1 progeny was allowed to self pollinate.

$$1+1+2 = 4$$

OR

If 1600 plants were obtained in F2 progeny, write the number of plants having traits :

- (i) Tall with round seeds
- (ii) Short with wrinkled seeds

Write the conclusion of the above experiment.



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